



Fig. 5 Graphic Procedure for Determining the Corrected Temperature Rise

### CALCULATIONS

The energy change,  $Q$ , measured in this calorimeter is calculated by multiplying the net corrected temperature change,  $\Delta T_c$ , by the energy equivalent,  $e$ , of the calorimeter and its contents.

$$Q = (\Delta T_c) (e)$$

If  $\Delta T_c$  is measured in degrees C and  $e$  is expressed in calories per degree C,  $Q$  will be reported in calories.

The energy equivalent,  $e$ , is determined by a calibration procedure described below under Standardization. The value  $e$  is simply the effective heat capacity of the calorimeter.

No factory standardization is currently recorded for this calorimeter. Your  $W$  values should be approximately  $535 \pm 10 \text{ Cal/}^\circ\text{C}$ .