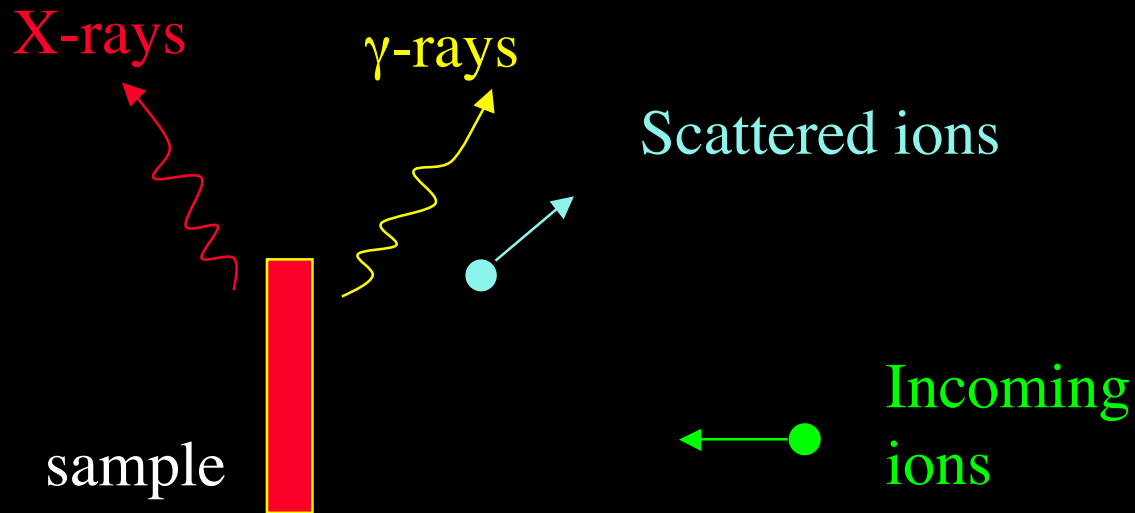


Fast ions

Different processes



The name of the **-technique-** depends on which “product” one uses as a probe.

Processes

Nuclear
and
Electron
Scattering
& Reactions

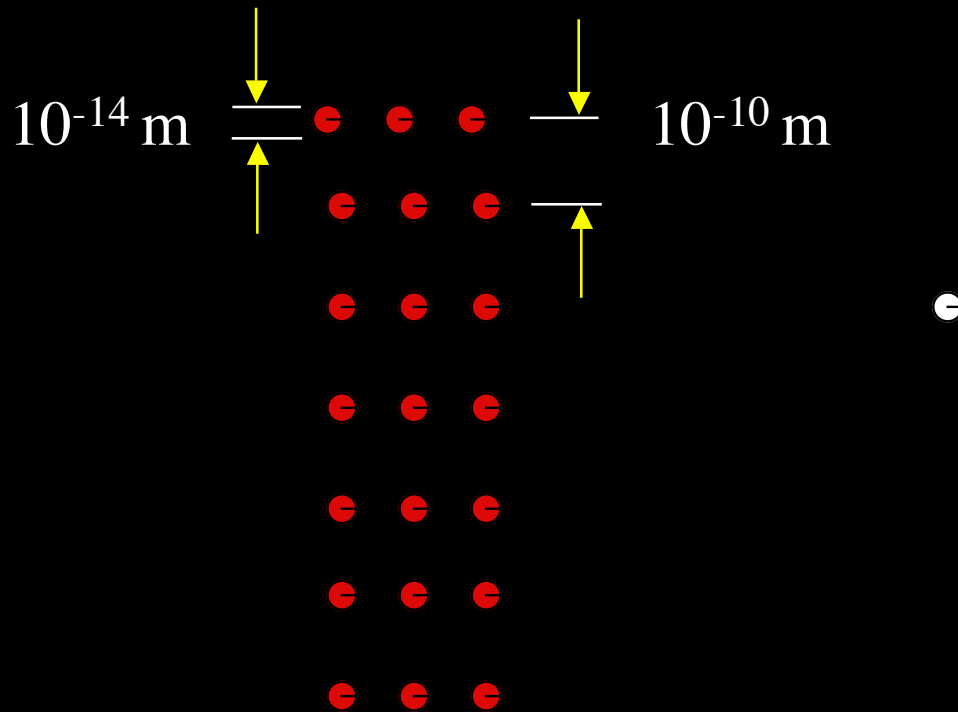
Speed
Charge state
Mass

Parameters

Mass

He - 4 u
Nuclei - 1 to 250 u
Electrons 0.00055 u

“Light” ion ($M_{\text{ion}} < M_{\text{target}}$)



Initial energy =
 E_{in}

CLASSICAL
MECHANICS

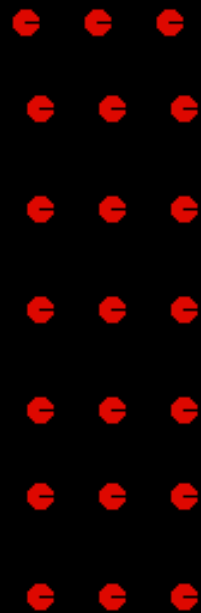
$$R_{\text{nuclei}} \approx 1.2 \cdot 10^{-15} A^{1/3} \text{ m}$$

mass target = 20

mass ion = 4

“Light” ion ($M_{\text{ion}} < M_{\text{target}}$)

Initial energy=
 E_{in}

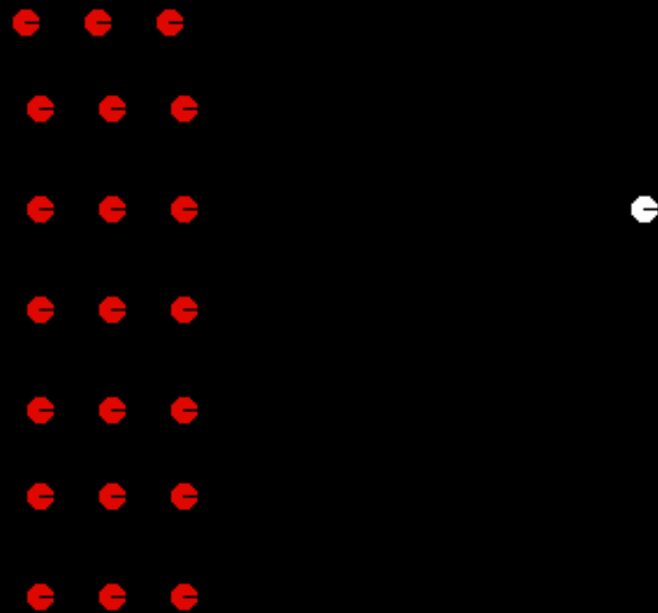


CLASSICAL
MECHANICS

mass target = 20

mass ion = 4

“Heavy” ion ($M_{\text{ion}} > M_{\text{target}}$)



mass target = 20

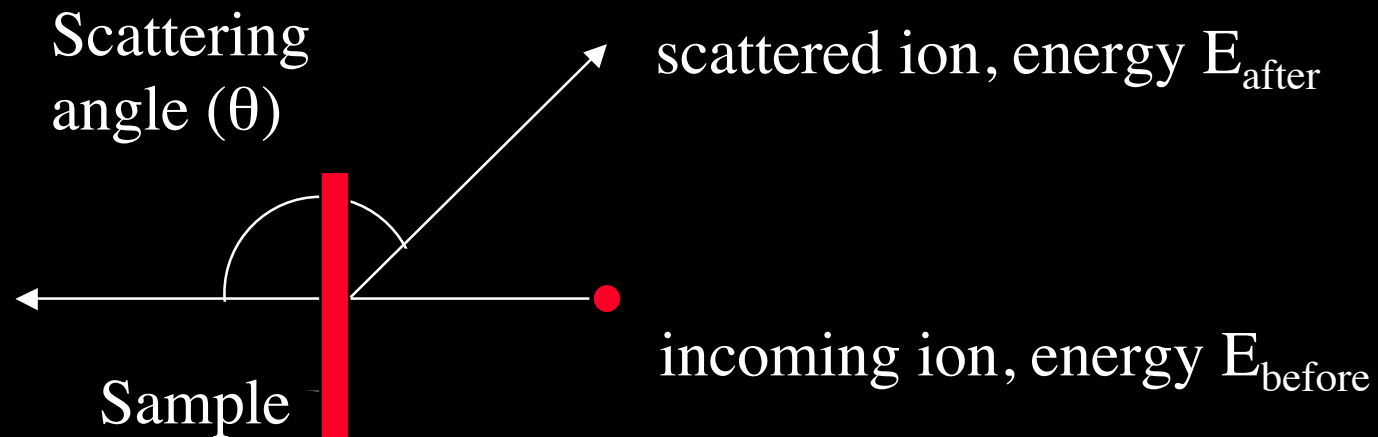
mass ion = 40

For a given scattering angle

Nuclear scattering

Scattering from a surface layer

$$E_{after} = k E_{before}$$



$$k = \left[\frac{\sqrt{M_{sample}^2 - M_{ion}^2 (\sin\theta)^2} + M_{ion} \cos(\theta)}{M_{ion} + M_{sample}} \right]^2$$

The energy of the ion

- ◆ The energy of the scattered ion can be calculated and measured EXACTLY!
- ◆ The probability of scattering depends on the energy of the incoming ions, for He below 2 MeV, we have the Rutherford region.

(Rutherford or Coulomb region => elastic nuclear collisions)

The probability of scattering

$$\frac{d\sigma}{d\Omega} \propto \left[\frac{Z_{ion} Z_{sample}}{E_{ion}} \right]^2$$

Electron and nuclear contributions to the scattering

